8E Combustion

Metal

Oxides

of Mass

Heating Zinc

Phlogiston

in Air

1. Burning Fuels		
	A chemical substance from which stored energy can be	
Fuel	transferred usefully to make	
	things happen.	
	Used in hydrogen-powered	
Fuel Cell	vehicles, releasing energy	
	from hydrogen.	
Fuel Cell Word	l Equation	
Hydrogen + ox	ygen → water	
Reactants	The starting substances- on	
reactants	left of word equation.	
Products	The new substances made-	
Troducts	on right of word equation.	
	Burning, usually in air. The	
	reaction gives out energy	
Combustion	which is transferred to the	
	surroundings by heating or	
	light.	
	Fuels formed from living	
Fossil Fuels	organisms that died millions	
	of years ago- petrol, diesel	
	Only contain carbon and	
Hydrocarbons	hydrogen atoms- petrol,	
	diesel	
Combustion	The carbon and hydrogen	
of	atoms react with oxygen.	
Hydrocarbons	The carbon reacts to form	
Try drocar bons	carbon dioxide.	
Carbon	Carbon dioxide will turn	
Dioxide	limewater cloudy.	

	reaction gives out energy which is transferred to the				
Combustion			3. Fire Safety		
	surroundings by heating or			A reaction that releases	
	light.		Exothermic	energy that we can feel as	
	Fuels formed from living			heat- combustion	
Fossil Fuels	organisms that died millions	Thermometer		Used to measure a change in	
	of years ago- petrol, diesel		inermometer	the temperature.	
	Only contain carbon and			Three factors allow	
Hydrocarbons	hydrogen atoms- <i>petrol</i> ,			combustion to	
	diesel			occur.	
Combustion	The carbon and hydrogen		Fire Triangle		
of	atoms react with oxygen.				
Hydrocarbons	The carbon reacts to form				
i iyu ocai bolis	carbon dioxide.			FUEL	
Carbon	Carbon dioxide will turn		Putting Out a	You must remove at least	
Dioxide	limewater cloudy.		Fire	one of the three factors.	
	2. Oxidation		<i>1</i> 11	Explosive	
0 ' 1 ' '				Heating may cause an	
	<u> </u>			explosion.	
CIXINE	Compound formed by oxidation.			Flammable	
				These substances catch fire	
				easily.	

	Oxidising
⟨♥⟩	These substances release
	oxygen.
Fire	Work by cooling a fire or
Extinguishers	stopping oxygen getting to
	the fuel.
Oil Fire	Water will sink through the
	oil and turn to steam making
	the fire spread out. Use
	foam or a fire blanket to
	keep oxygen away.
	Water conducts electricity
Electrical Fire	so you may get a serious
	shock. Turn off the
	electricity and use a powder
	or carbon dioxide
	extinguisher.

Formed when metals react

rearrange to form the products, no new atoms are made and none disappear.
Forms a white powder zinc oxide. The mass will appear

to increase because the zinc

If the product is a gas it may

burned that was then proven

the mass has decreased.
A substance scientists used to think explained why things

has combined with the

oxygen in air.

Gas Products escape and make it seem like

not to exist.

metal + oxygen → metal oxide Mass is never gained or lost in a chemical reaction. The

with oxygen.

Conservation atoms in reactants just

	extga.is.rer.
	4. Air Pollution
Complete	Carbon burns in plenty of air
Combustion	only forming carbon dioxide.
Incomplete	Not enough oxygen for all the
Combustion	carbon to react with.
	 carbon dioxide- linked to
D.,	global warming
Products of	carbon monoxide-
Incomplete	poisonous gas
Combustion	 soot- damage lungs and
	trigger asthma
lua a contiti a a	Small amounts of other
Impurities	substances in fuels.
Sulfur	Formed when hydrocarbons
Dioxide	have a sulfur impurity.
Nitrogen	Formed by high engine
Oxide	temperatures causing nitrogen
Oxide	and oxygen in air to react.
	Something that can harm
Pollutants	living things and damage the
	environment.
	Found in cars to react carbon
Catalytic	monoxide with more oxygen
Converter	forming carbon dioxide. Also
	breaks down nitrogen oxides.

	Sulfur dioxide and nitrogen
Acid Rain	oxides rise into the air and
Aciu Kaiii	dissolve in water vapour. The
	rain is now more acidic.
	Neutralisation reactions used
Controlling	to remove acidic gases from
Acid Rain	chimney smoke. Acidic soil
Acid Kain	/water can be neutralised by
	adding calcium carbonate.

5. Global Warming		
Greenhouse	Trap energy from the Sun in	
Gases	the atmosphere <i>e.g. carbon</i>	
Gases	dioxide	
	Energy trapped by	
Greenhouse	greenhouse gases is	
Effect	transferred back to the	
Ellect	Earth's surface causing it to	
	warm up.	
Earth's	The temperature of the Earth	
Temperature	has fluctuated over time it is	
Over Time	rising rapidly now though.	
	Increase in global	
Global	temperature due to more	
Warming	greenhouse gases in the air	
	and the greenhouse effect.	
	Resulting from global	
Climate	warming- changes to	
Change	weather patterns, more	
	storms, flood, droughts, etc.	
	There is now lots of evidence	
Evidence	for global warming. average	
LVIGETICE	temperatures are increasing	
	and ice caps are melting.	

Lesson	Memorised?
1. Burning Fuels	
2. Oxidation	
3. Fire Safety	
4. Air Pollution	
5. Global Warming	